Instruction Manual

HI 83215

Grow Master Basic for Nutrient Analyses





Dear Customer,

Thank you for choosing a Hanna product. Please read this instruction manual carefully before using the instrument. This manual will provide you with the necessary information for the correct use of the instrument. If you need additional technical information, do not hesitate to e-mail us at tech@hannainst.com.

TABLE OF CONTENTS		
PRELIMINARY EXAMINATION	2	
GENERAL DESCRIPTION		
ABBREVIATIONS		
SPECIFICATIONS		
PRECISION AND ACCURACY		
PRINCIPLE OF OPERATION		
FUNCTIONAL DESCRIPTION	<i>6</i>	
TIPS FOR AN ACCURATE MEASUREMENT	7	
HEALTH & SAFETY	10	
METHOD REFERENCE TABLE	11	
OPERATIONAL GUIDE	11	
SETUP	13	
HELP MODE	15	
INTRODUCTION	16	
SAMPLE PREPARATION	19	
AMMONIA HR	23	
AMMONIA MR		
AMMONIA LR	27	
NITRATE HR	29	
NITRATE MR	31	
NITRATE LR	33	
PHOSPHORUS HR	35	
PHOSPHORUS MR		
PHOSPHORUS LR		
POTASSIUM HR		
POTASSIUM MR		
POTASSIUM LR		
ERRORS AND WARNINGS		
DATA MANAGEMENT		
STANDARD METHODS		
ACCESSORIES		
WARRANTY		
HANNA LITEDATLIDE	51	

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PRELIMINARY EXAMINATION

Please examine this product carefully. Make sure that the instrument is not damaged. If any damage occurred during shipment, please notify your local Hanna Office.

Each Meter is supplied complete with:

- Four Sample Cuvettes and Caps
- Sample Preparation Kit (see page 19)
- Cloth for wiping cuvettes (1 pcs)
- Scissors
- AC/DC Power Adapter
- Instruction Manual

The sample preparation kit contains:

- 4 cuvettes (10 mL) with caps
- 2 plastic beakers (100 and 170 mL)
- 1 graduated cylinder (100 mL)
- 1 syringe with screw rim (60 mL)
- 1 syringe (5 mL)
- 1 funnel
- 25 filter discs
- 1 spoon
- 2 pipettes
- Carbon powder packets (50 pcs)
- 1 Demineralizer Bottle with filter cap for about 12 liters of deionized water (depending on the hardness level of water to be treated)

<u>Note</u>: Save all packing material until you are sure that the instrument works correctly. Any defective item must be returned in its original packing with the supplied accessories.

GENERAL DESCRIPTION

HI 83215 is a multiparameter bench photometer dedicated for Nutrient analyses. It can measure 12 different methods using specific liquid or powder reagents. The amount of reagent is precisely dosed to ensure maximum reproducibility.

HI 83215 bench photometer can be connected to a PC via an USB cable. The optional HI 92000 Windows® Compatible Software helps users manage all their results.

HI 83215 has a powerful interactive user support that assists the user during the analysis process.

Each step in the measurement process is help supported. A tutorial mode is available in the Setup Menu.

ABBREVIATIONS

°C: degree Celsius

°F: degree Fahrenheit

μg/L: micrograms per liter (ppb)
mg/L: milligrams per liter (ppm)

g/L: grams per liter (ppt)

mL: milliliter

HR: high range

MR: medium range

LR: low range

SPECIFICATIONS

Light Life Life of the instrument Light Detector Silicon Photocell

Environment 0 to 50° C (32 to 122° F);

max 95% RH non-condensing

Power Supply external 12 Vdc power adapter

built-in rechargeable battery

Dimensions 235 x 200 x 110 mm (9.2 x 7.87 x 4.33")

Weight 0.9 Kg

For specifications related to each method (e.g. range, resolution, etc.) refer to the related measurement section.

PRECISION AND ACCURACY

<u>Precision</u> is how closely repeated measurements agree with each other. Precision is usually expressed as standard deviation (SD).

 $\underline{\text{Accuracy}}$ is defined as the nearness of a test result to the true value.

Although good precision suggests good accuracy, precise results can be inaccurate. The figure explains these definitions.

For each method, the precision is expressed in the related measurement section.



Precise, accurate

Not precise, accurate





Precise, not accurate

Not precise, not accurate



PRINCIPLE OF OPERATION

Absorption of light is a typical phenomenon of interaction between electromagnetic radiation and matter. When a light beam crosses a substance, some of the radiation may be absorbed by atoms, molecules or crystal lattices.

If pure absorption occurs, the fraction of light absorbed depends both on the optical path length through the matter and on the physical-chemical characteristics of substance according to the Lambert-Beer Law:

Where:

-log I/I = Absorbance (A)

I = intensity of incident light beam I = intensity of light beam after absorption

 $\epsilon_{_{\!\lambda}} \; = \;$ molar extinction coefficient at wavelength λ

= molar concentration of the substance

d = optical path through the substance

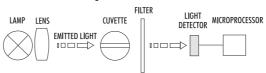
Therefore, the concentration "c" can be calculated from the absorbance of the substance as the other factors are known.

Photometric chemical analysis is based on the possibility to develop an absorbing compound from a specific chemical reaction between sample and reagents.

Given that the absorption of a compound strictly depends on the wavelength of the incident light beam, a narrow spectral bandwidth should be selected as well as a proper central wavelength to optimize measurements.

The optical system of HI 83215 is based on special subminiature tunasten lamps and narrow-band interference filters to guarantee both high performance and reliable results.

Three measuring channels allow a wide range of tests.



Instrument block diagram (optical layout)

A microprocessor controlled special tungsten lamp emits radiation which is first optically conditioned and beamed through the sample contained in the cuvette. The optical path is fixed by the diameter of the cuvette. Then the light is spectrally filtered to a narrow spectral bandwidth, to obtain a light beam of intensity I_{\circ} or I.

The photoelectric cell collects the radiation I that is not absorbed by the sample and converts it into an electric current, producing a potential in the mV range.

The microprocessor uses this potential to convert the incoming value into the desired measuring unit and to display it on the LCD.

The measurement process is carried out in two phases: first the meter is zeroed and then the actual measurement is performed.

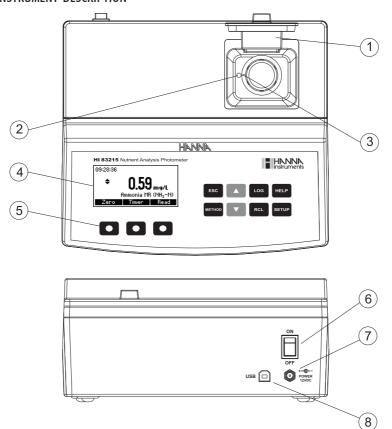
The cuvette has a very important role because it is an optical element and thus requires particular attention. It is important that both the measurement and the calibration (zeroing) cuvette are optically identical to provide the same measurement conditions. Most methods use the same cuvette for both, so it is important that measurements are taken at the same optical point. The instrument and the cuvette cap have special marks that must be aligned in order to obtain better reproducibility.

The surface of the cuvette must be clean and not scratched. This is to avoid measurement interference due to unwanted reflection and absorption of light. It is recommended not to touch the cuvette walls with hands

Furthermore, in order to maintain the same conditions during the zeroing and the measurement phases, it is necessary to cap the cuvette to prevent any contamination.

FUNCTIONAL DESCRIPTION

INSTRUMENT DESCRIPTION



- 1) Open Cuvette Lid
- 2) Indexing mark
- 3) Cuvette point
- 4) Liquid Crystal Display (LCD)
- 5) Splash proof keypad
- 6) ON/OFF power switch
- 7) Power input connector
- 8) USB connector

KEYPAD DESCRIPTION

The keypad contains 8 direct keys and 3 functional keys with the following functions:

Press to perform the function displayed above it on the LCD.

Press to exit the current screen.

Press to access the select method menu.

Press to move up in a menu or a help screen, to increment a set value, to access second level functions

Press to move down in a menu or a help screen, to decrement a set value, to access second level functions.

Log Press to log the current reading.

Press to recall the log.

Press to display the help screen.

SETUP Press to access the setup screen.

TIPS FOR AN ACCURATE MEASUREMENT

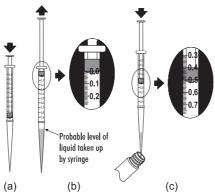
The instructions listed below should be carefully followed during testing to ensure most accurate results.

- Color or suspended matter in large amounts may cause interference, and should be removed by treatment with active carbon and filtration: refere to the SAMPLE PREPARATION Chapter (see page 19)
- Ensure the cuvette is filled correctly: the liquid in the cuvette forms a convexity on the top; the bottom of this convexity must be at the same level as the 10 mL mark.

COLLECTING AND MEASURING SAMPLES

In order to measure exactly 0.5 mL of reagent with the <u>1 mL syringe</u>:
 (a) push the plunger completely into the syringe and insert the tip into the solution.

- (b) pull the plunger up until the lower edge of the seal is exactly on the 0.0 mL mark.
- (c) take out the syringe and clean the outside of the syringe tip. Be sure that no drops are hanging on the tip of the syringe, if so eliminate them. Then, keeping the syringe in vertical position above the cuvette, push the plunger down into the syringe until the lower edge of the seal is exactly on the 0.5 mL mark. Now the exact amount of 0.5 mL has been added to the cuvette, even if the tip still contains some solution.

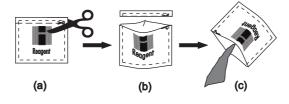


USING LIQUID AND POWDER REAGENTS

- Proper use of the dropper:
 - (a) for reproducible results, tap the dropper on the table for several times and wipe the outside of the dropper tip with a cloth.
 - (b) always keep the dropper bottle in a vertical position while dosing the reagent.

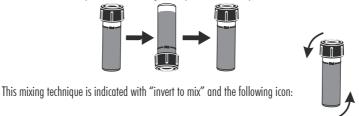


- Proper use of the powder reagent packet:
 - (a) use scissors to open the powder packet;
 - (b) push the edges of the packet to form a spout;
 - (c) pour out the content of the packet.



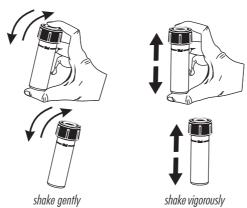
USING CUVETTES

- Proper mixing is very important for reproducibility of the measurements. The right way of mixing a
 cuvette is specified for each method in the related chapter.
 - (a) invert the cuvette a couple of times or for a specified time: hold the cuvette in the vertical position. Turn the cuvette upside-down and wait for all of the solution to flow to the cap end, then return the cuvette to the upright vertical position and wait for all of the solution to flow to the cuvette bottom. This is one inversion. The correct speed for this mixing technique is 10-15 complete inversions in 30 seconds.

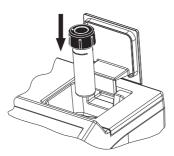


(b) shaking the cuvette, moving the cuvette up and down. The movement may be gentle or vigorous.

This mixing method is indicated with "shake gently" or "shake vigorously", and one of the following icons:



Pay attention to push the cuvette completely down in the holder and to align the white point on the
cap to the indexing mark on the meter.



- In order to avoid reagent leaking and to obtain more accurate measurements, close the cuvette first with the supplied HDPE plastic stopper and then the black cap.
- Whenever the cuvette is placed into the measurement cell, it must be dry outside, and free of fingerprints, oil or dirt. Wipe it thoroughly with HI 731318 or a lint-free cloth prior to insertion.
- Shaking the cuvette can generate bubbles in the sample, causing higher readings. To obtain accurate measurements, remove such bubbles by swirling or by gently tapping the cuvette.
- Do not let the reacted sample stand too long after reagent is added. For best
 accuracy, respect the timings described in each specific method.
- It is possible to take multiple readings in a row, but it is recommended to take
 a new zero reading for each sample and to use the same cuvette for zeroing
 and measurement when possible (for most precise results follow the measurement
 procedures carefully).
- Discard the sample immediately after the reading is taken, or the glass might become permanently stained.
- All the reaction times reported in this manual are at 25 °C (77 °F). In general, the reaction time should be increased for temperatures lower than 20 °C (68 °F), and decreased for temperatures higher than 25 °C (77 °F).

INTERFERENCES

 In the method measurement section the most common interferences that may be present in an average sample matrix have been reported. It may be that for a particular treatment process other compounds do interfere with the method of analysis.

HEALTH & SAFETY



- The chemicals contained in the reagent kits may be hazardous if improperly handled.
- Read the Material Safety Data Sheet (MSDS) before performing tests.
- <u>Safety equipment</u>: Wear suitable eye protection and clothing when required, and follow instructions carefully.
- <u>Reagent spills</u>: If a reagent spill occurs, wipe up immediately and rinse with plenty of water.
 If reagent contacts skin, rinse the affected area thoroughly with water. Avoid breathing released vapors.
- Waste disposal: for proper disposal of reagent kits and reacted samples, refer to the Material Safety Data Sheet (MSDS).





METHOD REFERENCE TABLE

Method	Method description	Page
1	Ammonia HR	23
2	Ammonia MR	25
3	Ammonia LR	27
4	Nitrate HR	29
5	Nitrate MR	31
6	Nitrate LR	33

Method	Method description	Page
7	Phosphorus HR	35
8	Phosphorus MR	37
9	Phosphorus LR	39
10	Potassium HR	41
11	Potassium MR	44
12	Potassium LR	46

OPERATIONAL GUIDE

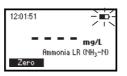
POWER CONNECTION AND BATTERY MANAGEMENT

The meter can be powered from an AC/DC adapter (included) or from the built-in rechargeable battery.

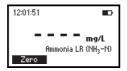
Note: Always turn the meter off before unplugging it to ensure no data is lost.

When the meter switches ON, it verifies if the power supply adapter is connected. The battery icon on the LCD will indicate the battery status:

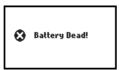
- battery is charging from external adapter



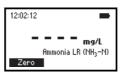
- battery capacity (no external adapter)



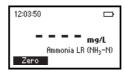
- battery Dead (no external adapter)



- battery fully charged (meter connected to AC/DC adapter)



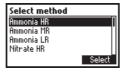
- battery Low (no external adapter)

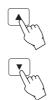


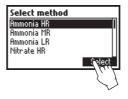
METHOD SELECTION

- Turn the instrument ON via the ON/OFF power switch.
- The meter will perform an autodiagnostic test. During this test, the Hanna Instrument logo will appear
 on the LCD. After 5 seconds, if the test was successful, the last method used will appear on the display.
- In order to select the desired method press the METHOD key and a screen with the available methods will appear.
- Press the AV keys to highlight the desired method. Press Select.









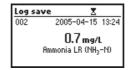
- After the desired method is selected, follow the measurement described in the related section.
- Before performing a test you should read all the instructions carefully.

DATA MANAGEMENT

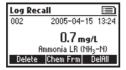
The instrument features a data log function to help you keep track of all your analysis. The data log can hold 200 individual measurements. Storing, viewing and deleting the data is possible using the LOG and RCL keys.

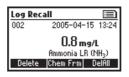
Storing data: You can store only a valid measurement. Press **LOG** and the last valid measurement will be stored with date and time stamps.





Viewing and deleting: You can view and delete the data log by pressing the RCL key. You can only delete the last saved measurement. Additionally, you can delete the data records all at once.

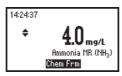






CHEMICAL FORM

Chemical form conversion factors are pre-programmed into the instrument and are method specific. In order to view the displayed result in the desired chemical form press \blacktriangle or \blacktriangledown to access the second level functions and then press the **Chem Frm** key to toggle between the available chemical forms for the selected method.





SETUP

In the Setup mode the instrument's parameters can be changed. Some parameters affect the measuring sequence and others are general parameters that change the behavior or appearance of the instrument.

Press **SETUP** to enter the setup mode.

Press ESC or SETUP to return to the main screen.

A list of setup parameters will be displayed with currently configured settings. Press **HELP** for additional information.

Press the \blacktriangle \blacktriangledown keys to select a parameter and change the value as follows:



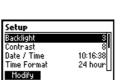
Backlight

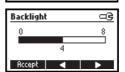
Values: 0 to 8.

Press the Modify key to access the backlight value.

Use the $\blacktriangleleft \blacktriangleright$ functional keys or the $\blacktriangle \blacktriangledown$ keys to increase or decrease the value.

Press the **Accept** key to confirm or **ESC** to return to the setup menu without saving the new value.





Contrast

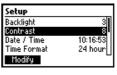
Values: 0 to 20.

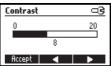
This option is used to set the display's contrast.

Press the **Modify** key to change the display's contrast.

Use the $\blacktriangleleft \blacktriangleright$ functional keys or the $\blacktriangle \blacktriangledown$ keys to increase or decrease the value.

Press the **Accept** key to confirm the value or **ESC** to return to the setup menu without saving the new value.





Date / Time

This option is used to set the instrument's date and time. Press the Modify key to change the date/time.

Press the

Functional keys to highlight the value to be modified (year, month, day, hour, minute or second). Use the
keys to change the value.

Press the **Accept** key to confirm or **ESC** to return to the setup without saving the new date or time.

Time format

Option: AM/PM or 24 hour.

Press the functional key to select the desired time format.

Date format

Press the **Modify** key to change the Date Format.

Use the **\(\bigcup \)** keys to select the desired format.

Press the **Accept** key to confirm or **ESC** to return to the setup menu without saving the new format.

Language

Press the corresponding key to change the language.

If the new language cannot be loaded, the previously selected language will be reloaded.

Tutorial

Option: Enable or Disable.

If enabled this option will provide the user short guide related to the current screen.

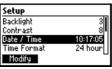
Press the functional key to enable/disable the tutorial mode.

Beeper

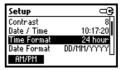
Option: Enable or Disable.

When enabled, a short beep is heard every time a key is pressed. A long beep alert sounds when the pressed key is not active or an error condition is detected.

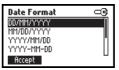
Press the functional key to enable/disable the beeper.



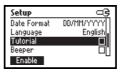


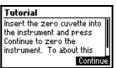


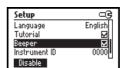












Instrument ID

Option: 0 to 9999.

This option is used to set the instrument's ID (identification number). The instrument ID is used while exchanging data with a PC.

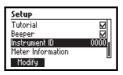
Press the **Modify** key to access the instrument ID screen. Press the **A V** keys in order to set the desired value.

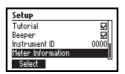
Press the **Accept** key to confirm the value or **ESC** to return to the setup menu without saving the new value.

Meter information

Press the **Select** key to view the instrument model, firmware version, language version and instrument serial number.

Press **ESC** to return to the Setup mode.







HELP MODE

HI 83215 offers an interactive contextual help mode that assists the user at any time.

To access the help screens press **HELP**.

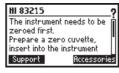
The instrument will display additional information related to the current screen. To read all the available information, scroll the text using the \blacktriangle \blacktriangledown keys.

Press the **Support** key to access a screen with Hanna service centers and their contact details.

Press the **Accessories** key to access a list of instrument reagents and accessories.

To exit support or accessories screens press **ESC** and the instrument will return to the previous help screen.

To exit help mode press the **HELP** or **ESC** key again and the meter will return to the previously selected screen.







INTRODUCTION

THE ROLE OF NUTRIENTS IN THE GROWTH AND PRODUCTION OF PLANTS

The three elements that are mostly needed by the plants are nitrogen (N), phosphorus (P) and potassium (K). They are called the macronutrients while other elements, needed by plants in smaller amounts, are called microelements. In hydroponics, plants need a balanced nutritive solution, composed of macro and microelements.

Shortage or excess of even only one nutritive element may cause an imbalance in plant physiology and in the absorption of the other nutrients. Nutrients shortages may result in irregular plant growth, low resistance to diseases, scarce production both in quantity and quality, while nutrients excess may cause waste of fertilizer, pollution of the groundwater and the possible accumulation of dangerous substances in the crops produced.

NITROGEN

Nitrogen (N) is mostly absorbed by plants as nitrates (NO_3^-) and, in smaller amount, in the form of ammonium (NH_4^+). In hydroponics, an adequate ratio between the two forms is generally used in nutritive solutions.

3010110113.	
PRESENT IN	proteins, enzymes, chlorophyll, hormones, vitamins, DNA and RNA
ACTION	• is fundamental for plants in phase of growth
	promotes the lengthening of trunks and sprouts
	• increases the production of foliage
	helps to absorb other nutrients
	(in particular phosphorus)
	assists a bigger production for both size and
	number of fruits
SHORTAGE	• slower growth
EFFECTS	• smaller leaves
	• yellowing of leaves
	• smaller fruits
	• premature ripening
EXCESS EFFECTS	reduction in resistance to diseases and atmospheric agents
	 increase of water demand (caused by an excessive production of leaves)
	bad quality of fruits
	delayed ripening
	reduction in potassium absorption

PHOSPHORUS

Phosphorus (P) has an important role in many fundamental biochemical and physiological processes. Plants take up phosphorus in the form of phosphate ion (PO $_4^{3-}$).

DDECENT IN	DNA IDNA ATD ADD
PRESENT IN	DNA and RNA, ATP, ADP
ACTION	 stimulates the roots growth
	• stimulates blooming
	stimulates fecundation and ripeness
	• strengthens the plant tissues
	• is necessary in the formation of seeds
SHORTAGE	• delayed ripening
<i>EFFECTS</i>	• slower growth
	• small leaves
	 decrease of production (smaller fruits and difficult seeds formation)
	• reduction of root system
EXCESS EFFECTS	premature ripening
	 excess of fruit-setting
	ullet negative effects on the absorption of some microelements such as iron, zinc, boron and copper

POTASSIUM

Potassium (K) is essential in proteic synthesis. The problem of lack of potassium is quite frequent in calcareous soils.

Potassium is absorbed as K^+ .

PRESENT IN	tissues responsible for the growth of plants (primary and secondary meristems), embryos and cell vacuole
ACTION	 improves the quality of fruits and flowers gives more resistance both to frost and to diseases caused by fungi (increases the cuticular thickness) regulates the cellular turgidity (helps to regulate the osmotic processes and increases the resistance to dryness) regulates the stomatic opening and closing (it means a strong influence on transpiration and photosynthesis)
SHORTAGE	• slower growth
<i>EFFECTS</i>	 smaller fruits, less colored and less preserved
	• increase of transpiration
	• less resistance to the cold
EXCESS EFFECTS	reduced absorption of calcium and magnesium increase of water consumption
	• increase of the substrate salinity

IRRIGATION WATER (LOW RANGE)

In agricultural areas it is quite common to find altered values in the chemical composition of irrigation waters. The problem concerns mostly the high nitrate concentration, usually determined by excessive fertilization or irrational liquid manure spreading. The analysis of irrigation waters allows us to find out which are the substances present in major or minor quantity and to organize an advantageous fertilization plan.

For example, if the quantity of water utilized for crop cultivation is 250 mm/ha (= 2500000 L/ha) and the nitrate (NO $_3$ ⁻) concentration is 150 mg/L (34 mg/L as nitrate-nitrogen NO $_3$ -N), soil receives 85 kg/ha of nitrogen. In choosing type and quantity of fertilizer to be used, it is important to consider this information, in order not to waste fertilizer nor to induce soil pollution.

NUTRIENTS SOLUTIONS (MEDIUM AND HIGH RANGE)

The nutrients requirements of the plants are determined by the type of plant, its age and the environmental conditions. The control of chemical composition of nutrients solutions given to the plants is an operation that allows a correct preparation of the fertilizer. In analyzing the solution, it is possible to choose between medium range and high range values, depending on the concentration of substances.

Medium range usually covers the analysis of residual solutions in recycling systems. The nutritive elements are differently absorbed by the plants, hence the nutrient solution loses substances, becomes impoverished and must be enriched.

High range normally corresponds to the typical values of nutrients solutions. It is therefore possible to verify that the solution given to the plants contains the correct quantities of nutritive substances.

SAMPLE PREPARATION

SAMPLE PREPARATION PROCEDURE

<u>Note</u>: If the water sample to be analyzed is very turbid, let it stand in a beaker for a while until most of the solid particles have settled. Then, use the pipette to transfer the supernatant solution to the other beaker and prepare the samples as described below. To prevent the displacement of the settled solids at the bottom of the beaker, do not induce air bubbles into the solution.

IRRIGATION WATER (LR):

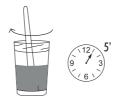
• Measure 100 mL of sample with the graduated cylinder.



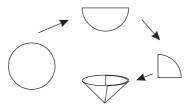
• If the solution contains some turbidity or color, pour it in the large 170 mL beaker and add a powder packet of active carbon.



• Mix well using the spoon and then wait for 5 minutes.



• Fold a filter disc twice as shown in the figure. Separate one side from the other three to form a cone. Insert the folded filter disc in the funnel.



Filter the treated sample into an empty beaker.
 The sample is now ready.



Note: Filter at least 40 mL of solution if all the four methods will be tested. If the solution is still turbid or colored, treat it again with a packet of active carbon. After use, throw the filter disc away and wash the syringe and the filter assembly well. Always use a new disc for another sample.

NUTRIENTS SOLUTION (MR):

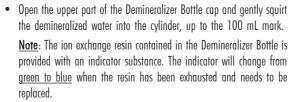
• Use the graduated cylinder to measure exactly 20 mL of sample.



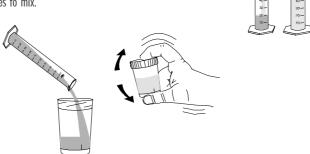
• Remove the cap and fill the Demineralizer Bottle with tap water.







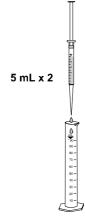




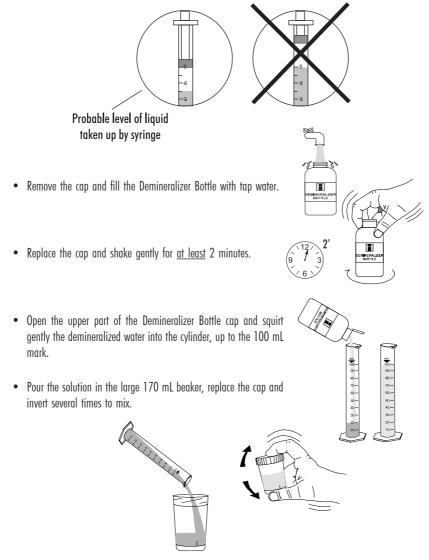
• If the solution contains some turbidity or color, add a powder packet of active carbon and follow the procedure described for Irrigation Water (LR).

NUTRIENTS SOLUTION (HR):

 Add 10 mL of sample to the graduated cylinder using the 5 mL syringe (twice).



Note: To measure exactly 5 mL of sample with the syringe, push the plunger completely into the syringe and insert the tip into the sample. Pull the plunger out until the lower edge of the seal is on the 5 mL mark of the syringe.



If the solution contains some turbidity or color, add a powder packet of active carbon and follow the
procedure described for Irrigation Water (LR).

AMMONIA HIGH RANGE

SPECIFICATIONS

Range 0 to 100 mg/L Resolution 1 mg/L

Accuracy $\pm 1 \text{ mg/L} \pm 4\% \text{ of reading at 25 °C}$

Typical EMC $\pm 1 \text{ mg/L}$

Deviation

Light Source Tungsten lamp with narrow band interference filter @ 420 nm

Method Adaptation of the ASTM Manual of Water and Environmental Technology, D1426-92,

Nessler method. The reaction between ammonia and reagents causes a yellow tint in

the sample.

REQUIRED REAGENTS

CodeDescriptionQuantityHI 93715A-0First Reagent4 drops (6 drops for seawater)HI 93715B-0Second Reagent4 drops (10 drops for seawater)

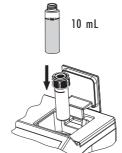
REAGENT SETS

HI 93715-01 Reagents for 100 tests HI 93715-03 Reagents for 300 tests For other accessories see page 50.

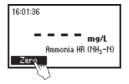
MEASUREMENT PROCEDURE

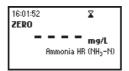
 $\underline{\text{Note}}$: for sample preparation follow the NUTRIENTS SOLUTION (HR) procedure at page 21.

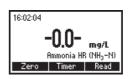
- Select the *Ammonia HR* method using the procedure described in the *Method Selection* section (see page 12).
- Fill the cuvette with 10 mL of unreacted sample (up to the mark) and replace the cap.
- Place the cuvette into the holder and close the lid.



 Press the Zero key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.

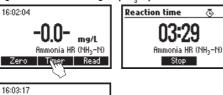


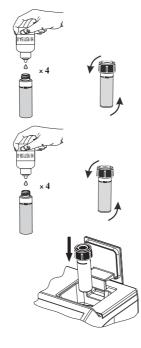


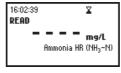


Ammonia HR

- · Remove the cuvette.
- Add 4 drops of HI 93715A-0 First Reagent (6 drops for seawater analysis). Replace the cap and mix the solution by inverting the cuvette a couple of times.
- Add 4 drops of HI 93715B-O Second Reagent (10 drops for seawater analysis). Replace the cap and mix the solution by inverting the cuvette a couple of times.
- · Reinsert the cuvette into the instrument.
- Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait for 3 minutes and 30 seconds and press Read. When the timer ends the meter will perform the reading. The instrument displays the results in mg/L of ammonia nitrogen (NH₃-N).

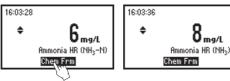


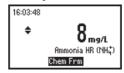






- Press ▲ or ▼ to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of ammonia (NH₃) and ammonium (NH₄+).





• Press **\(\)** or **\(\)** to return to the measurement screen.

INTERFERENCES

Interference may be caused by: acetone, alcohols, aldehydes, glycine, hardness above 1 g/L, iron, organic chloramines, sulfide, various aliphatic and aromatic amines.

AMMONIA MEDIUM RANGE

SPECIFICATIONS

 $\textbf{Range} \hspace{1.5cm} \textbf{0.0 to 50.0 mg/L}$

Resolution 0.5 mg/L

Accuracy ± 0.5 mg/L $\pm 4\%$ of reading at 25 °C

Typical EMC $\pm 0.5 \text{ mg/L}$

Deviation

Light Source Tungsten lamp with narrow band interference filter @ 420 nm

Method Adaptation of the ASTM Manual of Water and Environmental Technology, D1426-92,

Nessler method. The reaction between ammonia and reagents causes a yellow tint in

the sample.

REQUIRED REAGENTS

CodeDescriptionQuantityHI 93715A-0First Reagent4 drops (6 drops for seawater)HI 93715B-0Second Reagent4 drops (10 drops for seawater)

REAGENT SETS

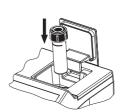
HI 93715-01 Reagents for 100 tests HI 93715-03 Reagents for 300 tests For other accessories see page 50.

MEASUREMENT PROCEDURE

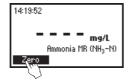
<u>Note</u>: for sample preparation follow the NUTRIENTS SOLUTION (MR) procedure at page 20.

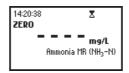
- Select the *Ammonia MR* method using the procedure described in the *Method Selection* section (see page 12).
- Fill the cuvette with 10 mL of unreacted sample (up to the mark) and replace the cap.
- Place the cuvette into the holder and close the lid.





 Press the Zero key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.



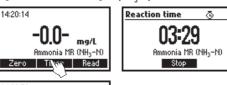


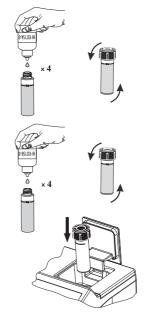
25

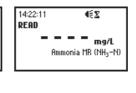


Ammonia MR

- · Remove the cuvette.
- Add 4 drops of HI 93715A-0 First Reagent (6 drops for seawater analysis). Replace the cap and mix the solution by inverting the cuvette a couple of times.
- Add 4 drops of HI 93715B-O Second Reagent (10 drops for seawater analysis). Replace the cap and mix the solution by inverting the cuvette a couple of times.
- Reinsert the cuvette into the instrument.
- Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait for 3 minutes and 30 seconds and press Read. When the timer ends the meter will perform the reading. The instrument displays the results in mg/L of ammonia nitrogen (NH₂-N).









- Press ▲ or ▼ to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of ammonia (NH₃) and ammonium (NH₄+).



Press
 or
 to return to the measurement screen.

INTERFERENCES

Interference may be caused by:

acetone, alcohols, aldehydes, glycine, hardness above 1 g/L, iron, organic chloramines, sulfide, various aliphatic and aromatic amines.

AMMONIA LOW RANGE

SPECIFICATIONS

 $\textbf{Range} \hspace{1cm} 0.0 \hspace{1mm} \text{to} \hspace{1mm} 10.0 \hspace{1mm} \text{mg/L}$

Resolution 0.1 mg/L

Accuracy ± 0.1 mg/L $\pm 4\%$ of reading at 25 °C

Typical EMC $\pm 0.1 \text{ mg/L}$

Deviation

Light Source Tungsten lamp with narrow band interference filter @ 420 nm

Method Adaptation of the ASTM Manual of Water and Environmental Technology, D1426-92,

Nessler method. The reaction between ammonia and reagents causes a yellow tint in

the sample.

REQUIRED REAGENTS

CodeDescriptionQuantityHI 93715A-0First Reagent4 drops (6 drops for seawater)HI 93715B-0Second Reagent4 drops (10 drops for seawater)

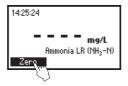
REAGENT SETS

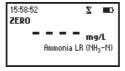
HI 93715-01 Reagents for 100 tests HI 93715-03 Reagents for 300 tests For other accessories see page 50.

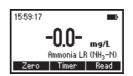
MEASUREMENT PROCEDURE

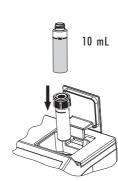
 $\underline{\text{Note}}$: for sample preparation follow the IRRIGATION WATER (LR) procedure at page 19.

- Select the *Ammonia LR* method using the procedure described in the *Method Selection* section (see page 12).
- Fill the cuvette with 10 mL of unreacted sample (up to the mark) and replace the cap.
- Place the cuvette into the holder and close the lid.
- Press the Zero key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.



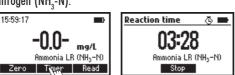


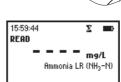


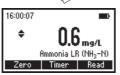


27 Ammonia LR

- · Remove the cuvette.
- Add 4 drops of HI 93715A-0 First Reagent (6 drops for seawater analysis). Replace the cap and mix the solution by inverting the cuvette a couple of times.
- Add 4 drops of HI 93715B-O Second Reagent (10 drops for seawater analysis). Replace the cap and mix the solution by inverting the cuvette a couple of times.
- Reinsert the cuvette into the instrument.
- Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait for 3 minutes and 30 seconds and press Read. When the timer ends the meter will perform the reading. The instrument displays the results in mg/L of ammonia nitrogen (NH₂-N).





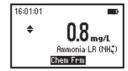


15:59:17

- Press
 or
 to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of ammonia (NH_a) and ammonium (NH₄+).







Press
 or
 to return to the measurement screen.

INTERFERENCES

Interference may be caused by: acetone, alcohols, aldehydes, glycine, hardness above 1 g/L, iron, organic chloramines, sulfide, various aliphatic and aromatic amines.

Ammonia LR

NITRATE HIGH RANGE

SPECIFICATIONS

Range 0 to 300 mg/L Resolution 5 mg/L

Accuracy \pm 10 mg/L \pm 8% of reading at 25 °C

Typical EMC $\pm 5 \text{ mg/L}$

Deviation

Light Source Tungsten lamp with narrow band interference filter @ 525 nm

Method Adaptation of the cadmium reduction method. The reaction between nitrate-nitrogen

and the reagent causes an amber tint in the sample.

REQUIRED REAGENTS

CodeDescriptionQuantityHI 93728-0Powder reagent1 packet

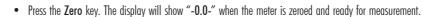
REAGENT SETS

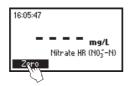
HI 93728-01 Reagents for 100 tests **HI 93728-03** Reagents for 300 tests For other accessories see page 50.

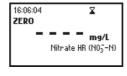
MEASUREMENT PROCEDURE

Note: for sample preparation follow the NUTRIENTS SOLUTION (HR) procedure at page 21.

- Select the *Nitrate HR* method using the procedure described in the *Method Selection* section (see page 12).
- Using the pipette, fill the cuvette with 6 ml of sample, up to half of its height, and replace the cap.
- Place the cuvette into the holder and close the lid.







 Remove the cuvette and add the content of one packet of HI 93728-0 reagent.



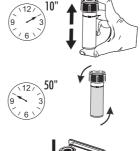
٥

6 mL

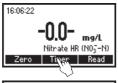


Nitrate HR

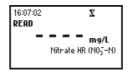
- Replace the cap and immediately shake vigorously up and down for exactly 10 seconds. Continue to mix by inverting the cuvette gently for 50 seconds, while taking care not to induce air bubbles. Powder will not completely dissolve. Time and way of shaking could sensitively affect the measurement.
- Reinsert the cuvette into the instrument, taking care not to shake it.
- Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait for 4 minutes and 30 seconds and press Read. When the timer ends the meter will perform the reading. The instrument displays the results in mg/L of nitrate-nitrogen.

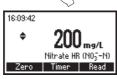




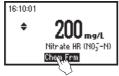








- Press ▲ or ▼ to access the second level functions.
- Press the **Chem Frm** key to convert the result in mg/L of nitrate (NO₂⁻).





• Press lacktriangle or lacktriangle to return to the measurement screen.

INTERFERENCES

Interference may be caused by:

Ammonia and amines, as urea and primary aliphatic amines

Chloride above 100 ppm

Chlorine above 2 ppm

Copper

Iron(III)

Strong oxidizing and reducing substances

Sulfide must be absent

Nitrate HR

NITRATE MEDIUM RANGE

SPECIFICATIONS

Range 0 to 150 mg/L **Resolution** 2.5 mg/L

Accuracy $\pm 5 \text{ mg/L} \pm 8\% \text{ of reading at } 25 ^{\circ}\text{C}$

Typical EMC ± 2.5 mg/L

Deviation

Light Source Tungsten lamp with narrow band interference filter @ 525 nm

Method Adaptation of the cadmium reduction method. The reaction between nitrate-nitrogen

and the reagent causes an amber tint in the sample.

REQUIRED REAGENTS

CodeDescriptionQuantityHI 93728-0Powder reagent1 packet

REAGENT SETS

HI 93728-01 Reagents for 100 tests **HI 93728-03** Reagents for 300 tests For other accessories see page 50.

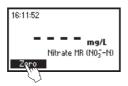
MEASUREMENT PROCEDURE

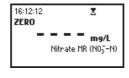
Note: for sample preparation follow the NUTRIENTS SOLUTION (MR) procedure at page 20.

- Select the Nitrate MR method using the procedure described in the Method Selection section (see page 12).
- Using the pipette, fill the cuvette with 6 ml of sample, up to half of its height, and replace the cap.
- Place the cuvette into the holder and close the lid.



• Press the Zero key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.





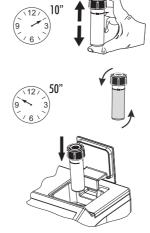


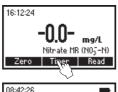
 Remove the cuvette and add the content of one packet of HI 93728-0 reagent.



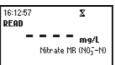
Nitrate MR

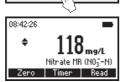
- Replace the cap and immediately shake vigorously up and down for exactly 10 seconds. Continue to mix by inverting the cuvette gently for 50 seconds, while taking care not to induce air bubbles. Powder will not completely dissolve. Time and way of shaking could sensitively affect the measurement.
- Reinsert the cuvette into the instrument, taking care not to shake it.
- Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait for 4 minutes and 30 seconds and press Read. When the timer ends the meter will perform the reading. The instrument displays the results in mg/L of nitrate-nitrogen.











- Press ▲ or ▼ to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of nitrate (NO₂-).





• Press lacktriangle or lacktriangle to return to the measurement screen.

INTERFERENCES

Interference may be caused by:

Ammonia and amines, as urea and primary aliphatic amines

Chloride above 100 ppm

Chlorine above 2 ppm

Copper

Iron(III)

Strong oxidizing and reducing substances

Sulfide must be absent

Nitrate MR

NITRATE LOW RANGE

SPECIFICATIONS

 $\textbf{Range} \hspace{1.5cm} \textbf{0.0 to 30.0 mg/L}$

Resolution 0.5 mg/L

Accuracy $\pm 1.0 \text{ mg/L} \pm 8\% \text{ of reading at 25 °C}$

Typical EMC ± 0.5 mg/L

Deviation

Light Source Tungsten lamp with narrow band interference filter @ 525 nm

Method Adaptation of the cadmium reduction method. The reaction between nitrate-nitrogen

and the reagent causes an amber tint in the sample.

REQUIRED REAGENTS

CodeDescriptionQuantityHI 93728-0Powder reagent1 packet

REAGENT SETS

HI 93728-01 Reagents for 100 tests HI 93728-03 Reagents for 300 tests For other accessories see page 50.

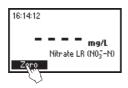
MEASUREMENT PROCEDURE

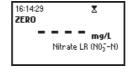
Note: for sample preparation follow the IRRIGATION WATER (LR) procedure at page 19.

- Select the Nitrate LR method using the procedure described in the Method Selection section (see page 12).
- Using the pipette, fill the cuvette with 6 ml of sample, up to half of its height, and replace the cap.
- Place the cuvette into the holder and close the lid.



• Press the **Zero** key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.





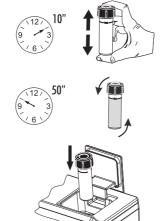
16:14:47

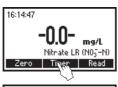
-0.0- mg/L
Nitrate LR (N03-N)
Zero Timer Read

 Remove the cuvette and add the content of one packet of HI 93728-0 reagent.

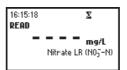
Nitrate LR

- Replace the cap and immediately shake vigorously up and down for exactly 10 seconds. Continue to mix by inverting the cuvette gently for 50 seconds, while taking care not to induce air bubbles. Powder will not completely dissolve. Time and way of shaking could sensitively affect the measurement.
- Reinsert the cuvette into the instrument, taking care not to shake it.
- Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait for 4 minutes and 30 seconds and press Read. When the timer ends the meter will perform the reading. The instrument displays the results in mg/L of nitrate-nitrogen.



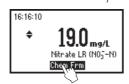


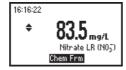






- Press ▲ or ▼ to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of nitrate (NO₂-).





• Press lacktriangle or lacktriangle to return to the measurement screen.

INTERFERENCES

Interference may be caused by:

Ammonia and amines, as urea and primary aliphatic amines

Chloride above 100 ppm

Chlorine above 2 ppm

Copper

Iron(III)

Strong oxidizing and reducing substances

Sulfide must be absent

Nitrate LR

PHOSPHORUS HIGH RANGE

SPECIFICATIONS

 Range
 0 to 100 mg/L

 Resolution
 1 mg/L

Accuracy $\pm 5 \text{ mg/L} \pm 4\% \text{ of reading at 25 °C}$

Typical EMC Dev. $\pm 1 \text{ mg/L}$

Light Source Tungsten lamp with narrow band interference filter @ 525 nm

Method Adaptation of the Standard Methods for the Examination of Water and Wastewater,

18th edition, Amino Acid method. The reaction between phosphorus and reagents

causes a blue tint in the sample.

REQUIRED REAGENTS

CodeDescriptionQuantityHI 93706A-0Molybdate10 dropsHI 93706B-0Amino Acid Powder1 packet

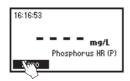
REAGENT SETS

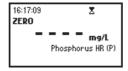
HI 93706-01 Reagents for 100 tests HI 93706-03 Reagents for 300 tests For other accessories see page 50.

MEASUREMENT PROCEDURE

Note: for sample preparation follow the NUTRIENTS SOLUTION (HR) procedure at page 21.

- Select the *Phosphorus HR* method using the procedure described in the *Method Selection* section (see page 12).
- Fill the cuvette with 10 mL of unreacted sample (up to the mark) and replace the cap.
- Place the cuvette into the holder and close the lid.
- Press the Zero key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.





- Remove the cuvette.
- Add 10 drops of HI 93706A-0 Molybdate reagent.





10 mL

Phosphorus HR

 Add the content of one packet of HI 93706B-0 Phosphorus Reagent B (Amino Acid) to the cuvette. Replace the cap and shake gently until completely dissolved.

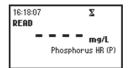


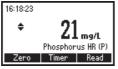
· Reinsert the cuvette into the instrument.

Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait
for 5 minutes and press Read. When the timer ends the meter will perform the reading. The instrument
displays the results in mg/L of phosphorus (P).

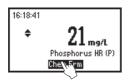


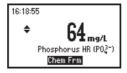


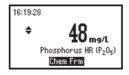




- Press lacktriangle or lacktriangle to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of phosphate (P0 $_4^{3-}$) and phosphorus pentoxide (P $_2^{0}$ 0 $_5$).







• Press lacktriangle or lacktriangle to return to the measurement screen.

INTERFERENCES

Interference may be caused by:
Sulfide
Chloride above 150000 mg/L
Calcium above 10000 mg/L as CaCO₃
Magnesium above 40000 mg/L as CaCO₃
Ferrous iron above 100 mg/L
Phosphorus HR

PHOSPHORUS MEDIUM RANGE

SPECIFICATIONS

Range 0.0 to 50.0 mg/L **Resolution** 0.5 mg/L

Accuracy $\pm 2.5 \text{ mg/L} \pm 4\% \text{ of reading at } 25 \,^{\circ}\text{C}$

Typical EMC Dev. ± 0.5 mg/L

Light Source Tungsten lamp with narrow band interference filter @ 525 nm

Method Adaptation of the Standard Methods for the Examination of Water and Wastewater,

 18^{h} edition, Amino Acid method. The reaction between phosphorus and reagents

causes a blue tint in the sample.

REQUIRED REAGENTS

CodeDescriptionQuantityHI 93706A-0Molybdate10 dropsHI 93706B-0Amino Acid Powder1 packet

REAGENT SETS

HI 93706-01 Reagents for 100 tests **HI 93706-03** Reagents for 300 tests For other accessories see page 50.

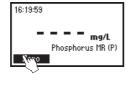
MEASUREMENT PROCEDURE

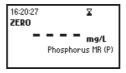
Note: for sample preparation follow the NUTRIENTS SOLUTION (MR) procedure at page 20.

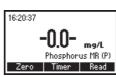
- Select the *Phosphorus* method using the procedure described in the *Method Selection* section (see page 12).
- Fill the cuvette with 10 mL of unreacted sample (up to the mark) and replace the cap.
- Place the cuvette into the holder and close the lid.



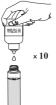
Press the Zero key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.







- · Remove the cuvette.
- Add 10 drops of HI 93706A-0 Molybdate reagent.



Phosphorus MR

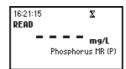
- Add the content of one packet of HI 93706B-0 Phosphorus Reagent B (Amino Acid) to the cuvette. Replace the cap and shake gently until completely dissolved.
- · Reinsert the cuvette into the instrument.

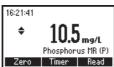


Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait
for 5 minutes and press Read. When the timer ends the meter will perform the reading. The instrument
displays the results in mg/L of phosphorus (P).

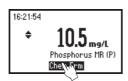




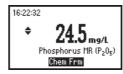




- Press ▲ or ▼ to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of phosphate (P0 $_4^{3-}$) and phosphorus pentoxide (P $_2$ 0 $_z$).







• Press lacktriangle or lacktriangle to return to the measurement screen.

INTERFERENCES

Interference may be caused by:

Sulfide

Chloride above 150000 mg/L

Calcium above 10000 mg/L as CaCO,

Magnesium above 40000 mg/L as CaCO₂

Ferrous iron above 100 mg/L

Phosphorus MR

PHOSPHORUS LOW RANGE

SPECIFICATIONS

Range 0.0 to 10.0 mg/L **Resolution** 0.1 mg/L

Accuracy ± 0.5 mg/L $\pm 4\%$ of reading at 25 °C

Typical EMC Dev. ± 0.1 mg/L

Light Source Tungsten lamp with narrow band interference filter @ 525 nm

Method Adaptation of the Standard Methods for the Examination of Water and Wastewater,

 18^{h} edition, Amino Acid method. The reaction between phosphorus and reagents

causes a blue tint in the sample.

REQUIRED REAGENTS

CodeDescriptionQuantityHI 93706A-0Molybdate10 dropsHI 93706B-0Amino Acid Powder1 packet

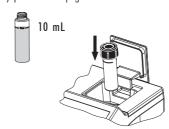
REAGENT SETS

HI 93706-01 Reagents for 100 tests **HI 93706-03** Reagents for 300 tests For other accessories see page 50.

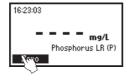
MEASUREMENT PROCEDURE

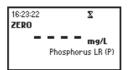
Note: for sample preparation follow the IRRIGATION WATER (LR) procedure at page 19.

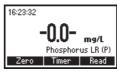
- Select the *Phosphorus* method using the procedure described in the *Method Selection* section (see page 12).
- Fill the cuvette with 10 mL of unreacted sample (up to the mark) and replace the cap.
- Place the cuvette into the holder and close the lid.



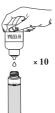
• Press the Zero key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.





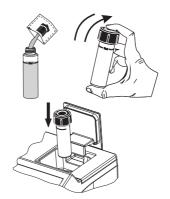


- · Remove the cuvette.
- Add 10 drops of HI 93706A-0 Molybdate reagent.



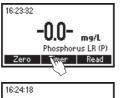
Phosphorus LR

 Add the content of one packet of HI 93706B-0 Phosphorus Reagent B (Amino Acid) to the cuvette. Replace the cap and shake gently until completely dissolved.

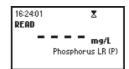


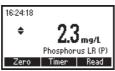
Reinsert the cuvette into the instrument.

Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait
for 5 minutes and press Read. When the timer ends the meter will perform the reading. The instrument
displays the results in mg/L of phosphorus (P).

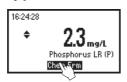


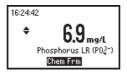


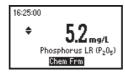




- Press \blacktriangle or \blacktriangledown to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of phosphate (P_{2}^{3}) and phosphorus pentoxide (P_{2}^{0}).







Press
 or
 to return to the measurement screen.

INTERFERENCES

Interference may be caused by:
Sulfide
Chloride above 150000 mg/L
Calcium above 10000 mg/L as CaCO₃
Magnesium above 40000 mg/L as CaCO₃
Ferrous iron above 100 mg/L

POTASSIUM HIGH RANGE

SPECIFICATIONS

Range 20 to 200 mg/L

Resolution 5 mg/L

Accuracy $\pm 30 \text{ mg/L} \pm 7\% \text{ of reading at 25 °C}$

Typical EMC $\pm 5 \text{ mg/L}$

Deviation

Light Source Tungsten lamp with narrow band interference filter @ 610 nm

Method Adaptation of the Turbidimetric Tetraphenylborate method. The reaction between

Potassium and reagents causes turbidity in the sample.

REQUIRED REAGENTS

<u>Code</u>	<u>Description</u>	Quantity		
HI 93750 A -0	Potassium Reagent	6 drops		
HI 93750 B -0	Powder Reagent	1 packet		

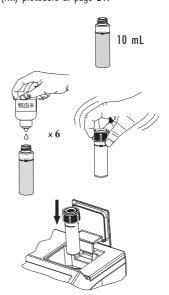
REAGENT SETS

HI 93750-01 Reagents for 100 tests **HI 93750-03** Reagents for 300 tests For other accessories see page 50.

MEASUREMENT PROCEDURE

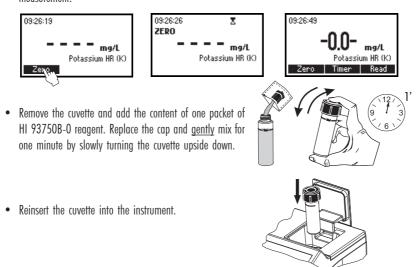
Note: for sample preparation follow the NUTRIENT SOLUTIONS (HR) procedure at page 21.

- Select the *Potassium HR* method using the procedure described in the *Method Selection* section (see page 12).
- Fill the cuvette with 10 mL of sample, up to the mark.
- Add six drops of HI 93750A-0, replace the cap and swirl the solution.
- Place the cuvette into the holder and close the lid.



41 Potassium HR

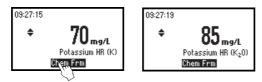
 Press the Zero key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.



Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait
for 2 minutes and press Read. When the timer ends the meter will perform the reading. The instrument
displays the results in mg/L (ppm) of potasium (K).



- Press ▲ or ▼ to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of potassium oxide (K,O)



- Press ▲ or ▼ to return to the measurement screen.
- For ULTRA HIGH RANGE samples: follow the procedure described at page 43.

INTERFERENCES

Interferences may be caused by:
Ammonium above 10 ppm
Calcium above 10000 ppm as CaCO₃
Chloride above 12000 ppm
Magnesium above 8000 ppm as CaCO₃
Sodium above 8000 ppm

POTASSIUM ULTRA HIGH RANGE

For samples containing more than 200 ppm of Potassium: follow the sample preparation procedure described at page 21 for NUTRIENTS SOLUTION (HR). Then add to the graduated cylinder 20 mL of the prepared sample (for HR) and fill the cylinder with demineralized water from the Demineralizer Bottle up to the 100 mL mark.

Follow the MEASUREMENT PROCEDURE at page 41.

Read the result in mg/L of Potassium on the display and multiply the reading by 5 to obtain the actual concentration of Potassium.

43 Potassium HR

POTASSIUM MEDIUM RANGE

SPECIFICATIONS

Range 10 to 100 mg/L Resolution 2.5 mg/L

Accuracy $\pm 15 \text{ mg/L} \pm 7\% \text{ of reading at } 25 ^{\circ}\text{C}$

Typical EMC $\pm 2.5 \text{ mg/L}$

Deviation

Light Source Tungsten lamp with narrow band interference filter @ 610 nm

Method Adaptation of the Turbidimetric Tetraphenylborate method. The reaction between

Potassium and reagents causes turbidity in the sample.

REQUIRED REAGENTS

<u>Code</u>	<u>Description</u>	Quantity		
HI 93750 A -0	Potassium Reagent	6 drops		
HI 93750 B -0	Powder Reggent	1 nacket		

REAGENT SETS

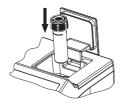
HI 93750-01 Reagents for 100 tests HI 93750-03 Reagents for 300 tests For other accessories see page 50.

MEASUREMENT PROCEDURE

Note: for sample preparation follow the NUTRIENTS SOLUTION (MR) procedure at page 20.

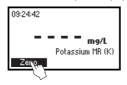
- Select the *Potassium MR* method using the procedure described in the *Method Selection* section (see page 12).
- Fill the cuvette with 10 mL of sample, up to the mark.
- Add six drops of HI 93750A-0, replace the cap and swirl the solution.
- Place the cuvette into the holder and close the lid.

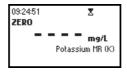


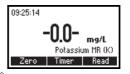


Potassium MR

• Press the **Zero** key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.

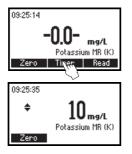




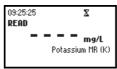


- Remove the cuvette and add the content of one packet of HI 93750B-O reagent. Replace the cap and <u>gently</u> mix for one minute by slowly turning the cuvette upside down.
- · Reinsert the cuvette into the instrument.
- Press Timer and the display will show the countdown prior
 to the measurement or, alternatively, wait for 2 minutes
 and press Read. When the timer ends the meter will
 perform the reading. The instrument displays the results in
 mg/L (ppm) of potassium (K).

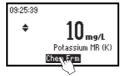


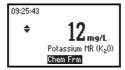






- Press ▲ or ▼ to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of potassium oxide (K_20) .





• Press **\(\)** or **\(\)** to return to the measurement screen.

INTERFERENCES

Interferences may be caused by:
Ammonium above 10 ppm
Calcium above 10000 ppm as CaCO₃
Chloride above 12000 ppm
Magnesium above 8000 ppm as CaCO₃
Sodium above 8000 ppm

POTASSIUM LOW RANGE

SPECIFICATIONS

Range 0.0 to 20.0 mg/L **Resolution** 0.5 mg/L

Accuracy $\pm 3.0 \text{ mg/L} \pm 7\% \text{ of reading at } 25 \,^{\circ}\text{C}$

Typical EMC $\pm 0.5 \text{ mg/L}$

Deviation

Light Source Tungsten lamp with narrow band interference filter @ 610 nm

Method Adaptation of the Turbidimetric Tetraphenylborate method. The reaction between

Potassium and reagents causes turbidity in the sample.

REQUIRED REAGENTS

<u>Code</u>	<u>Description</u>	Quantity		
HI 93750 A -0	Potassium Reagent	6 drops		
HI 93750 B -0	Powder Reagent	1 packet		

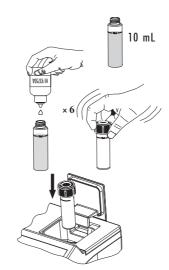
REAGENT SETS

HI 93750-01 Reagents for 100 tests **HI 93750-03** Reagents for 300 tests For other accessories see page 50.

MEASUREMENT PROCEDURE

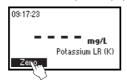
Note: for sample preparation follow the IRRIGATION WATER (LR) procedure at page 19.

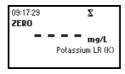
- Select the Potassium LR method using the procedure described in the Method Selection section (see page 12).
- Fill the cuvette with 10 mL of sample, up to the mark.
- Add 6 drops of HI 93750A-0 Potassium Reagent, replace the cap and swirl the solution.
- Place the cuvette into the holder and close the lid.

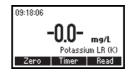


Potassium LR 46

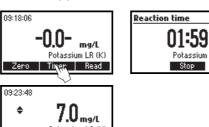
• Press the **Zero** key. The display will show "-0.0-" when the meter is zeroed and ready for measurement.

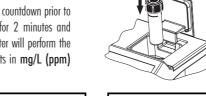


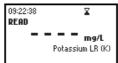




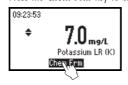
- Remove the cuvette and add the content of one packet of HI 93750B-O reagent. Replace the cap and <u>gently</u> mix for one minute by slowly turning the cuvette upside down.
- Reinsert the cuvette into the instrument.
- Press Timer and the display will show the countdown prior to the measurement or, alternatively, wait for 2 minutes and press Read. When the timer ends the meter will perform the reading. The instrument displays the results in mg/L (ppm) of potassium (K).

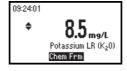






- Press ▲ or ▼ to access the second level functions.
- Press the Chem Frm key to convert the result in mg/L of potassium oxide (K,0).





Press ▲ or ▼ to return to the measurement screen.

INTERFERENCES

Interferences may be caused by:
Ammonium above 10 ppm
Calcium above 10000 ppm as CaCO₃
Chloride above 12000 ppm
Magnesium above 8000 ppm as CaCO₃
Sodium above 8000 ppm

ERRORS AND WARNINGS

The instrument shows clear warning messages when erroneous conditions appear and when measured values are outside the expected range. These messages are described bellow.



No Light: The light source is not functioning properly.



Light Leak: There is an excess amount of ambient light reaching the detector



Inverted cuvettes: The sample and the zero cuvettes are inverted.



Battery Low: The battery capacity is lower than 10%.



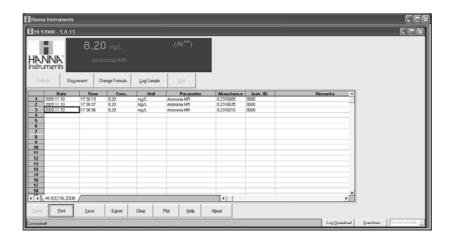
Light Low: The instrument cannot adjust the light level. Please check that the sample does not contain any debris.



Light High: There is too much light to perform a measurement. Please check the preparation of the zero cuvette.

DATA MANAGEMENT

The analyzed data can be managed using Hanna's product HI92000, Windows® Compatible Software.



STANDARD METHODS			
<u>Description</u>	<u>Range</u>	<u>Method</u>	
Ammonia HR	0 to 100 mg/L	Nessler	
Ammonia MR	0.0 to 50.0 mg/L	Nessler	
Ammonia LR	0.0 to 10.0 mg/L	Nessler	
Nitrate HR	0 to 300 mg/L	Cadmium Reduction	
Nitrate MR	0 to 150 mg/L	Cadmium Reduction	
Nitrate LR	0.0 to 30.0 mg/L	Cadmium Reduction	
Phosphorus HR	0 to 100 mg/L	Amino Acid	
Phosphorus MR	0.0 to 50.0 mg/L	Amino Acid	
Phosphorus LR	0.0 to 10.0 mg/L	Amino Acid	
Potassium HR	20 to 200 mg/L	Turbidimetric	
Potassium MR	10 to 100 mg/L	Turbidimetric	
Potassium LR	0.0 to 20.0 mg/L	Turbidimetric	

 $\mbox{Windows}^{\mbox{\tiny{\circledR}}}$ is registered Trademark of "Microsoft Co."

ACCESSORIES

REAGENT SETS

100 phosphorus tests HI 93706-01 HI 93706-03 300 phosphorus tests HI 93715-01 100 ammonia tests 300 ammonia tests HI 93715-03 HI 93728-01 100 nitrate tests 300 nitrate tests HI 93728-03 HI 93750-01 100 potassium tests HI 93750-03 300 potassium tests

OTHER ACCESSORIES

HI 93703-55

HI 731318 cloth for wiping cuvettes (4 pcs) HI 731321 glass cuvettes (4 pcs) HI 731325W new cap for cuvette (4 pcs) HI 740034 cap for 100 mL beaker (6 pcs) HI 740036 100 mL plastic beaker (6 pcs) HI 740157 plastic refilling pipette (20 pcs) HI 740223 170 mL plastic beaker HI 740224 170 mL plastic beakers (12 pcs) HI 740225 60 mL graduated syringe 5 mL graduated syringe HI 740226 filter assembly HI 740227 HI 740228 filter discs (25 pcs) HI 740229 100 mL graduated cylinder HI 740230 230 mL demineralized water HI 92000 Windows compatible software HI 920013 PC connection cable HI 93703-50 cuvette cleaning solution (230 mL) HI 93703-54 dried resin (100 g)

activated carbon (50 pcs)

WARRANTY

All Hanna Instruments meters are warranted for two years against defects in workmanship and materials when used for its intended purpose and maintained according to the instructions.

This warranty is limited to repair or replacement free of charge.

Damages due to accident, misuse, tampering or lack of prescribed maintenance are not covered.

If service is required, contact your dealer. If under warranty, report the model number, date of purchase, serial number and the nature of the failure. If the repair is not covered by the warranty, you will be notified of the charges incurred.

If the instrument is to be returned to Hanna Instruments, first obtain a Returned Goods Authorization Number from the Customer Service Department and then send it with shipment costs prepaid. When shipping any instrument, make sure it is properly packaged for complete protection.

To validate your warranty, fill out and return the enclosed warranty card within 14 days from the date of purchase.

Recommendations for Users

Before using these products, make sure that they are entirely suitable for your specific application and for the environment in which they are used.

Operation of these instruments may cause unacceptable interferences to other electronic equipments, this requiring the operator to take all necessary steps to correct interferences.

Any variation introduced by the user to the supplied equipment may degrade the instruments' EMC performance.

To avoid damages or burns, do not put the instrument in microwave ovens. For yours and the instrument safety do not use or store the instrument in hazardous environments.

Hanna Instruments reserves the right to modify the design, construction and appearance of its products without advance notice.

HANNA LITERATURE

Hanna publishes a wide range of catalogs and handbooks for an equally wide range of applications. The reference literature currently covers areas such as:

- Water Treatment
- Process
- Swimming Pools
- Agriculture
- Food
- Laboratory

and many others. New reference material is constantly being added to the library.

For these and other catalogs, handbooks and leaflets contact your dealer or the Hanna Customer Service Center nearest to you. To find the Hanna Office in your vicinity, check our home page at www.hannainst.com.



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